GALVANIC CORROSION

In choosing a KOENIG EXPANDER[®] you must consider that the material of the sealing plug and the material of the production piece can show different electrical potentials. In the presence of an electrolyte (e.g. 5% water-NaCl solution), this potential difference causes electrochemical attack on the least noble of the metals in contact – galvanic corrosion. In this case, either the base material or its surface protection will become the anode and will be transferred to the pure metal of the cathode. The corrosion speed or the current density will be determined by the relative surface area or volume of the anode and cathode as illustrated below.



Large Anode Area —> Low Current Density at the Anode —> Slow Corrosion



Small Anode Area —> High Current Density at the Anode —> Fast Corrosion

GALVANIC CORROSION

EFFECT OF GALVANIC CORROSION

The following table shows the expected galvanic corrosion behavior of KOENIG EXPANDER[®] plugs in common base materials allowing for the relative surface areas of both metals, which influences the speed of corrosion.

Installation Material		Series									
		MB 600	MB 700	MB 850	CV 173	CV 588	SK/SKC	LK 600	LK 950	BF/BR	RE
Steel, Carbon/Low Alloy, Plain											
Steel, Carbon/Low, Zn Plated, Chromate											
Steel, Carbon/Low Alloy, Phosphatized											
Nitrided or Case Hardening Steel		Behavior Depends on the Method Used									
Stainless Steel, X8CrNiS18-9, 1.4305, ANSI 303											
Stainlesss Steel, X12CrS13, 1.4005, ANSI 416											
Cast Iron, EN 1561, Plain											
Cast Iron, EN 1561, Zn Plated, Chromate											
Cast Iron, EN 1561, Phosphatized											
Ductile Cast Iron, EN 1563, Plain											
Ductile Cast Iron, EN 1563, Zn Plated, Chromate											
Ductile Cast Iron, EN 1563, Phosphatized											
AlMg1SiCu	EN AW-6061										
AIMgSiPb	EN AW-6062										
AlCu4Mg1	EN AW-2024										
AlZnMgCu1,5	EN AW-7075										
G-AlSi7Mg	A-Norm 356										
G-AlSi9Mg											
G-AISi10Mg											
= Accelerated = Not Accelerate		ted	= Sli	ghtly Acc	elerated						

Key to the galvanic corrosion behavior of KOENIG EXPANDER[®] plugs in the presence of an electrolytic medium installed in base materials per the above table.

Suggestions to Prevent Galvanic Corrosion

- Choose materials with no or low potential difference.
- Use corrosion reducing designs, i.e. if possible prevent the accumulation of fluids on the outer surface of the workpiece.
- By using suitable surface coatings, corrosion attack can be considerably reduced.

Salt spray testing per DIN EN ISO 9227 is available upon request